

Etude des cinétiques d'absorption gaz liquide dans le cas de supports innovants

Doctorant·e

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Abstract

This thesis explores the kinetics of gas-liquid absorption using innovative supports, focusing on deep eutectic solvents (DESs) based on choline chloride (ChCl). The primary objective is to enhance the efficiency of carbon dioxide (CO₂) capture. Traditional amine-based solvents are compared with DESs, highlighting DESs' advantages such as low cost, low volatility, and high biodegradability. The research involves synthesizing and characterizing various ChCl-based solvents, including ChCl-glycerol and ChCl-potassium glycinate mixtures. Experimental studies on thermophysical properties—density, viscosity, refractive index, and gas solubility—reveal that increasing the water content in DESs reduces viscosity, thereby enhancing CO₂ solubility. The study also investigates the optimal ChCl concentration to balance reaction kinetics and diffusivity for maximum CO₂ absorption. Results indicate that while high ionic strength in DESs improves reaction rates, it also increases viscosity, potentially impeding CO₂ diffusion. This research offers valuable insights for developing efficient CO₂ capture technologies, emphasizing the importance of optimizing thermal and compositional parameters.